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**TFRIA** 









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### Introduction





# Which technique to observe what?

- Positon annihilation spectroscopy (PAS)
  - <u>Principle</u>: annihilation characteristics of positons into a sample (depend of electronic density and electron momentum distribution)
  - <u>Raw data</u>: Positon lifetime spectrum (PALS) and energy of the photons resulting from annihilation (CDB)
  - Microstructural information:
    - Size and density of vacancy-type defects (from mono vacancies to nanovoids)
    - Chemical environment of annihilation sites
- Small Angle neutron Scattering (SANS)
  - <u>Principle</u>: Scattering of mono-energetic neutron beam by a sample (submitted to magnetic field)
  - Raw data: Scattered intensity as a function of scattering angle (// and  $\perp$  to applied magnetic field)
  - <u>Microstructural information</u>: Size distribution, volume fraction, range of chemical composition of scattering features (solute clusters...)

### Which technique to observe what?



- □ Atom Probe Tomography (APT) Field Ion Microscopy (FIM)
  - Principle: field evaporation of a small sample
  - <u>Raw data</u>: position and chemical nature (APT) of atoms in the analysed volume (→ 3D reconstruction)
  - Microstructural information:
    - shape, size distribution, number density, chemical composition of solute clusters, precipitates, segregation on crystalline defects (APT)
    - Size distribution, number density of small vacancy clusters (FIM)
- Transmission Electron Microscopy (TEM)
  - <u>Principle</u>: Interaction between incident electron beam and a thin sample
  - <u>Raw data</u>: Images in transmission of the sample (2D), spectra containing chemical information (EDS, EELS)
  - Microstructural information:
    - Size distribution, number density of dislocation loops, voids and precipitates
    - Grain size
    - chemical mapping (EDS, EELS), information about composition of different phases, intergranular segregation

### Ex: Genesis platform





Zeiss – XB 540 SEM (EBSD, EDS, FIB)



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Jeol ARM 200F (double corrected, EELS, EDS, ADF, HAADF...)



Specific environment for radioactive materials







Cameca LEAP 4000X HR



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### Field Ion Microscopy

#### □ Sample:

- in the shape of a thin needle (< 50nm),
- cooled down cryogenic temperature (20-80K),
- under low pressure (~10<sup>-5</sup>mbar) of image gas (H, He, Ne),

microcanaux

ionisé

Ecran Galettes de phosphorescent

Système de

visualisation

• submitted to high voltage (1-11kV).

$$\Rightarrow E = \frac{V}{\beta R} \sim 10 - 50 \text{ V/nm}$$

#### □ Field ionization of the image gas

quelques

kilovolts

Echantillon

20 K à 80 K atome

0 0 35

#### Field evaporation of the surface atoms

#### $\Rightarrow$ Image of the atoms at the surface









 $G = \frac{L}{(m+1)R}$ 

 $G \sim 10^6 - 10^7$ 

### 3D – Field Ion Microscopy



Increment in depth from the number of evaporated plans



### Atom Probe Tomography

- Same principle than FIM but
  - No image gas (P~10<sup>-11</sup>mbar)
    - $\rightarrow$  Field evaporation of surface atoms
  - Position sensitive detector
     → Localisation of atom position
  - Time of flight mass spectrometry

15nm

→ Identification of the chemical nature



#### → 3D reconstruction of the analysed volume, at atomic scale



A. Etienne, n-irradiated 16MND5

### APT – tof mass spectrometry

- Assuming the whole potential energy if converted into kinetic energy :
- Evaporation pulse (laser or electric) is applied to the sample
   measurement of time-of-flight between pulse and arrival on detector





 $\frac{m}{n} = 2eV\frac{t_f^2}{L^2} = kV\frac{t_f^2}{T^2}$ 

Results are collected on a mass spectrum

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### APT-3D reconstruction



Position sensitive detector + inverse projection → Position of each atom at the surface



#### **Magnification**

$$G = \frac{L}{(m+1)R}$$

 $G\sim 10^6 - 10^7$ 

For each detected atom, the reconstructed volume is incremented by  $V_{at}/Q$ 

Knowing the analysed surface, it allows to reconstructed the 3<sup>rd</sup> dimension



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5nm

### APT – n-irradiated RPV steel





A. Etienne, SOTERIA, WP2 (RPV steel irradiated a high flux)

#### □ Solute clusters:

15.0 mm

- composition: 2.6%Mn 3.8%Si 6.3%Ni 5.3%Cu (Fe)
- number density: 3.7x10<sup>23</sup> m<sup>-3</sup>
- size distribution



#### □ Segregation along dislocations:

- composition: 0.95%Mn 1.1%Si 2.4%Ni 0.31%Cu 3.9P (Fe)
- density: 1.1x10<sup>14</sup> m<sup>-2</sup>



# APT – GB and interfacial segregation



L. Zhang: P segregation at carbide-matrix interface in RPV weld



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#### 13

#### Limitations

 trajectory aberrations resulting from difference in evaporation field can induce bias atom positioning (shape of particles, chemical composition)



 $E_B < E_A$ 

#### Matrix: chemical composition

Visible features

#### Performance

excess...

- detection efficiency: 35 to 80%
- detection limit: down 10 at. ppm in best cases
- spatial resolution: ~ 0.2 nm in depth 0.5-1nm laterally

distribution, shape, 3D repartition, volume fraction

Analysed volume: ~50x50x200 nm<sup>3</sup>





### APT – Performances and limitations

• Solute clusters and precipitates (0.5 - 50 nm): composition, number density, size

• Segregation on crystal defects (dislocations, loops, GB...): density, composition, Gibbs

### Transmission electron microscopy



#### S. Rouland





#### Diaphragme condenseur l entilles condenseu Porte-objet goniométriqu Echantillor Diaphragme de sélection d'aire Lentilles intermédiaires Lentille projectrice Films Ecran photographique d 'observation

#### □ Sample:

- in the shape of a thin foil (<100nm in thickness),
- under secondary vacuum (~10<sup>-5</sup>-10<sup>-7</sup>mbar),
- illuminated by electron beam (several10<sup>th</sup> to 100<sup>th</sup> of keV).

#### Interaction electron-matter

- diffraction, mass-thickness, phase contrasts
- energy loss, X-ray emission
- Image in transmission that can reach atomic resolution
  - information about phases, grain size, crystal defects...
  - Chemical analysis possible with specific detectors





# Diffraction contrast





- A parallel beam of electrons can be diffracted by a family of reticular planes (hkl) in Bragg conditions
  - Electron non- diffracted converge to the center or the focal plane
  - Diffracted electrons by the same plane family converge to the same point of the focal plane.
    - → Diffraction pattern in the focal plane



Each electron coming from a same point of the sample converge on same point of image plane forming the image into image plane of the objective

# Diffraction contrast





- The contrast is obtained by mean of **contrast aperture** located in the focal plane of the objective. This aperture allows to select electrons from direct beam or from one of the diffracted beam.
  - Deficit or excess of electrons at the location of diffracting zones that appear more or less bright.



The area selection aperture is located in the image plane of objective lens. It allows to select electrons coming from a give, area.
 →selection of the imaged area

### TEM - Example



FP7 Longlife, D3.3 – RPV steel, n-irradiated at high dose (M. Mayoral)



### Mass - thickness contrast





- Elastic scattering of results from interaction of incident electrons with electronic cloud or atom nucleus.
- It increases with:
  - Atomic number (Z)
  - Thickness of the sample

Through areas with high atomic number, electrons are more diffused (Z contrast)



Scattered amplitude



# Scanning TEM



- The incident beam is focused on a point of the sample (probe)
- The probe (<1nm) is scanned on the sample
- For each location, different signals can be collected to form the image



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### Contrast in STEM



Contrast on image is obtained thanks to annular detectors



High Angle Annular Dark Field (HAADF) detector

→ Heavy elements appear bright (Z-contrast)

Low Angle Annular Dark Field (ADF) detector

→ crystal defects bright (diffraction contrast)

#### Bright Field (BF) detector

→ Crystal defects and areas rich in heavy elements appear dark

#### Example



#### O. Startev et al, SOTERIA WP3

HAADF image of RPV steel



#### ➔ PAGB, lath boundaries and carbides



318 SOTERIA Training School - September 2018 - Polytechnic University of Valencia

23

### Chemical analysis





→ chemical mapping – Chemical composition (in transmission)

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#### Example



P. Edmondson et al, Acta Materialia (2017)

STEM EDS on n – irradiated high Ni weld

STEM BF



→ correlation between crystalline et chemical information

# (S)TEM – Performances and limitation.

#### Visible features

- Crystalline defects (GB, dislocations, dislocation loops, voids...): Size distribution (> 1nm), number density
- Precipitates: Size distribution (> 1nm), number density, chemical information (2D)
- Segregation: chemical information

#### Performance

- detection limit: ~ 1% (EDS)
- spatial resolution: ~ 0.2 nm
- Analysed volume: <100nm in depth x several  $\mu m^2$



#### Limitations

• 2D image in transmission



# Electropolishing

Sample preparation

#### 

Needs of very thin samples for FIM, APT and TEM (foil or needle)

- "high" quantities of matter can be a limitation for radioactive samples
- No control on the sample exact location
- limited to metals



### Sample preparation



#### □ Focused Ion Beam milling:

- Sputtering of sample atoms using ion beam (FIB)
- Imaging using electron beam (SEM)







### Summary



Technique	Sample shape	Analysed volume	Crystal defects	Segregations	Precipitates Solute clusters
3D-FIM	thin needle	50x50x200nm <sup>3</sup>	<ul> <li>dislocation, GB visible</li> <li>voids (0.5-few nm):</li> <li>SD, ND</li> </ul>	/	- visible in some cases
ΑΡΤ	thin needle	50x50x200nm <sup>3</sup>	- visible if segregated	- shape, CC, ND	- SD (0.5-50nm), ND, CC, shape
(S)TEM	thin foil	20-100nm thick several μm <sup>2</sup>	- GB - dislocation (D) - loops, voids, SFT (ND, SD) if > 1nm	<ul> <li>intergranular</li> <li>segregation:</li> <li>concentration</li> <li>profile</li> </ul>	<ul> <li>If size &gt; 1nm</li> <li>2D chemical mapping</li> <li>crystalline structure</li> <li>information about CC</li> </ul>
PAS	well polished plate	0-0.5mm depth several mm <sup>2</sup>	<ul> <li>V-type (from mono-V to nm voids): Size and chemical env.</li> </ul>	/	/
SANS	bulk	cm <sup>3</sup>	/	/	- SD (1-100nm), f <sub>v</sub> , CC range

CC = chemical composition / SD = Size distribution / ND = number density / CC = chemical composition  $f_v$  = volume fraction



### Thank you for you attention



http://gpm.labos.univ-rouen.fr/ http://genesis.univ-rouen.fr/





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### Additional slides





# Positon Annihilation Spectroscopy



- Positons (emitted by the desintegration of a radioelement) go through the sample
- After thermalisation, positons diffuse into the material where they can:
  - be trapped by vacancy-type defects
  - Annihilate with electrons from the material
- Annihilation results in the emission
   2 photon in opposite directions



PALS





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#### Doppler shift probes the local electron momentum



Doppler shift, ∆E is proportional to electron momentum, pL
 → Depends of the chemical environment



### Small angle neutron scattering





Mono-energetic neutron beam goes through a bulk sample

Scattered intensity as a function of scattering angle contain information about size distribution, shape and volume fraction of features

$$I(\vec{q}) \approx f_p(\Delta \rho_{\text{nuc}}^2 + \Delta \rho_{\text{mag}}^2 \sin^2 \alpha) \left| F_p(R, \vec{q}) \right|^2 S(\vec{q})$$

A-ratio contains information about chemical composition of features